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Synthesis of 1,3-Thiazine Derivatives from 2-Iminocyclopentanedithiocarboxylic Acid¹⁾

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Several new 3-substituted and 3,3-disubstituted-1,3,4,5,6,7-hexahydro-1-thioxocyclopenta[d]-[1,3]thiazines were obtained by the reactions of 2-iminocyclopentanedithiocarboxylic acid with some aldehydes and ketones. Some intermediate compounds in these reactions such as 2-amino-2-[2-(piperonylidenamino)-1-cyclopenten-1-yl]thiocarbonylthiocyclopentanedithiocarboxylic acid, 2-amino-2-[2-(vanillylidenamino)-1-cyclopenten-1-yl]thiocarbonylthiocyclopentanedithiocarboxylic acid, and 2-amino-2-[2-(4-methoxybenzylidenamino)-1-cyclopenten-1-yl]thiocarbonylthiocyclopentanedithiocarboxylic acid could be isolated.

Several new 1,3-thiazine derivatives were synthesized by the reactions of 2-iminocyclopentane-dithiocarboxylic acid (I) with carbonyl compounds. Compound I was isolated as the intermediate during the course of synthesis of 1,3,4,5,6,7-hexahydro-1-thioxocyclopenta[d][1,3]thiazine - 3 - spirocyclopentane. The coupling reactions of I with ketones were partly reported.²⁾

The reactions in question were effected by acid or base, and 3-substituted and 3,3-disubstituted-1,3,4,5,6,7-hexahydro-1-thioxocyclopenta[d][1,3]-thiazines (III) were produced. Aliphatic and

alicyclic ketones, in the presence of morpholine,²⁾ easily reacted with I to give respective thiazine derivatives. However, aldehydes and acetophenone under the same conditions did not afford the expected products.

On treatment with sulfuric acid in place of morpholine, we have found that aldehydes such as piperonal, vanillin, anisaldehyde, acetaldehyde, salicylaldehyde, benzaldehyde, o-methoxy or phydroxybenzaldehyde and p-tolualdehyde as well as ketones such as acetophenone, methyl ethyl ketone and acetone reacted with I to afford the corresponding thiazine derivatives III. This paper deals with the results.

Evidence for the structures of IIIa—j was provided by comparison of their IR, UV and NMR spectra with those of known 1,2,5,6,7,8-hexahydro-4-thioxo-4H-3,1-benzo[d]thiazine-2-spirocyclohexane, 2-ethyl-2,4,5-trimethyl-2,3-dihydro-6-thioxo-6H-1,3-thiazine, and 3,3-dimethyl-1,3,4,5,6,7-hexahydro-1-thioxocyclopenta[d][1,3]thiazine.²⁻⁴) The doublet near δ 6.15 (J=4—6 Hz) was assigned to a C-3 proton, and thus structures

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²⁾ T. Takeshima, M. Yokoyama, T. Imamoto, M. Akano and H. Asaba, J. Org. Chem., 34, 730 (1969).

IIIa, R=H, R'=3,4(-OCH₂O-)C₆H₃ IIa, R=H, R'=3,4(-OCH₂O-)C₆H₃ IIIb, R=H, R'=3-CH₃O-4-HO-C₆H₃ IIb, R=H, R'=3-CH₃O-4-HO-C₆H₃ IIIc, R=H, R'=p-CH₃O-C₆H₄ IIc, R=H, R'=p-CH₃O-C₆H₄ IIId, R=H, R'=CH₃ IIIe, R=CH₃, R'=C₆H₅

$$\begin{split} & \text{IIIf, R=H, R'=o-HO-C}_6H_4\\ & \text{IIIg, R=H, R'=C}_6H_5\\ & \text{IIIh, R=H, R'=o-CH}_3\text{O-C}_6H_4\\ & \text{IIIi, R=H, R'=p-HO-C}_6H_4\\ & \text{IIIi, R=H, R'=p-CH}_3\text{-C}_6H_4\\ & \text{IIIi, R=H, R'=p-CH}_3\text{-C}_6H_4\\ & \text{IIIk, R=CH}_3, R'=\text{C}_2H_5\\ & \text{IIII, R=R'=CH}_3 \end{split}$$

III were proposed as the corresponding thiazine structures.

It was shown that the reaction of methyl ethyl ketone with I was abortive.²⁾ However, the present experiment revealed that 3-ethyl-3-methyl-1,3,4,5,6,7 - hexahydro-1-thioxocyclopenta[d][1, 3]-thiazine IIIk, had a molecule of water of crystallization which was lost on heating at 138°C under reduced pressure and thus was of a different physical nature. Likewise, the thiazine derivatives IIIa and IIIj which were obtained by the reactions of I with piperonal and p-tolualdehyde, respectively, also had a molecule of water of crystallization which was lost on heating at 110°C under reduced pressure.

Under the acidic conditions, piperonal, vanillin and anisaldehyde gave respective intermediate compounds IIa, IIb and IIc at low temperature which changed to IIIa—b on treatment with sulfuric acid. Compound IIc was resistant to conversion into IIIc.

Structures IIa, IIb and IIc were assigned by spectral data (see Experimental).

Experimental

2-Amino-2-[2-(piperonylidenamino)-1 - cyclopenten-1-yl]thiocarbonylthiocyclopentanedithiocarboxylic acid (IIa). A solution of I (4.1 g), piperonal

(4 g) and sulfuric acid (96%, 1 g) in 150 ml of methanol was allowed to stand for 5 min at room temperature. The yellow material was collected, washed with methanol and recrystallized from methanol-N,N-dimethylformamide. Yellow long plates (IIa): yield 4.6 g, 70%; mp 211-214°C with dec. (slow heating), 215-220°C with dec. (rapid heating); UV \(\lambda_{\text{max}}^{\text{MeOH}} \) 311, 397 $m\mu$ (log $\varepsilon = 4.04$, 4.32); IR (KBr) 3380 (m, NH₂), 3296 (s, NH_2), 1609 (s, δNH_2), 1600 (vs, C=N), 1500 (s, ring), 1486 (s, ring), 1475 cm⁻¹ (m, ring); NMR (CF₃COOH) δ 9.00 (br, 1H, SH), 7.08 (m, 4H, C₆H₃ and CH=N), 6.30 (s, 2H, OCH₂O), 3.03 (m, 9H, C-1, $3,5,3',5' \underline{H}_2$, $2.20(m, 4H, C-4,4'\underline{H}_2)$; NMR (CD₃- $SOCD_3$) δ 10.80 (br, ca. 2H, $N\underline{H}_2$), 9.00 (br, ca. 1H, SH); Found: C, 53.56; H, 5.09; N, 6.18; S, 28.36%. Calcd for $C_{20}H_{22}N_2S_4O_2$: C, 53.30; H, 4.92; N, 6.22; S, 28.45%.

NMR absorptions of NH₂ and SH in CD₃SOCD₃ disappeared by addition of deuterium oxide.

Conversion of IIa into IIIa. A mixture of IIa (1 g), sulfuric acid (96%, 400 mg) and 50 ml of methanol was refluxed for ca. 1 hr until a clear solution was obtained. After cooling, a yellowish orange material was collected and recrystallized from N,N-dimethyl-formamide-water as yellowish orange long plates: mp 177—178°C, undepressed on admixture with an authentic specimen; yield 500 mg, 77%. IR and UV spectra of both specimens also were identical.

2-Amino-2- [2-(vanillylidenamino)-1-cyclopenten-1-yl]thiocarbonylthiocyclopentanedithiocarboxylic acid (IIb). To an ice-cold solution of I (4.7 g) and vanillin (5 g) in methanol (130 ml), sulfuric acid (96%, 2 g) was added and the solution was kept at 0°C for 5 min. The resulting yellow material was recrystallized from acetone-water. Yellow long plates (IIb): yield 5.2 g, 77%; mp 190—192°C with dec. (slow heat-

³⁾ T. Takeshima, T. Hayashi, M. Muraoka and T. Matsuoka, *ibid.*, **32**, 980 (1967).

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ing), 199—201°C with dec. (rapid heating); UV $\lambda_{\text{max}}^{\text{mooh}}$ 308, 397 m μ (log ε =4.13, 4.30); IR (CHCl₃) 3320 (m, NH₂), 3260 (m, NH₂), 1614 (s, δ NH₂), 1603 (vs, C=N), 1510 (m, ring), 1500 (m, ring), 1472 cm⁻¹ (m, ring); NMR (CF₃COOH) δ 8.97 (br, 1H, SH), 7.30 (m, 4H, C₆H₃ and CH=N), 6.12 (s, 1H, OH), 4.04 (s, 3H, OCH₃), 3.08 (m, 9H, C-1,3,5,3',5' H₂), 2.23 (m, 4H, C-4,4' H₂); NMR (CD₃SOCD₃) δ 10.86 (br, ϵa . 2H, NH₂), 8.97 (br, ϵa . 1H, SH); Found: C, 53.13; H, 5.49; N, 6.16; S, 28.53%. Calcd for C₂₀H₂₄N₂-S₄O₂: C, 53.07; H, 5.35; N, 6.19; S, 28.32%.

Conversion of IIb into IIIb. A mixture of IIb (1 g), sulfuric acid (96%, 0.5 g), N,N-dimethylform-amide (25 ml) and methanol (50 ml) was refluxed for 15 min and allowed to stand at room temperature for 2 hr and then was added water. The resulting material was recrystallized from N,N-dimethylform-amide-water as orange prisms: yield 400 mg, 62%; mp 229—238°C with dec., undepressed by addition of an authentic specimen. UV and IR spectra of both specimens also were identical.

2 - Amino - 2 - [2-(4-methoxybenzylidenamino)-1cyclopenten-1-yl]thiocarbonylthiocyclopentanedithiocarboxylic acid (IIc). To an ice-cold solution of I (2.5 g) and anisaldehyde (2.5 g) in methanol (100 ml), sulfuric acid (96%, 1 g) was added and kept at 0°C for 10 min. The reaction mixture was worked up as above. Yellow long plates (IIc): yield 3 g, 88%; mp 209-214°C with dec. (slow heating), 213-223°C with dec. (rapid heating); UV \(\lambda_{\text{max}}^{\text{MeOH}} \) 275, 392 $m\mu$ (log $\varepsilon = 4.15, 4.27$); IR (KBr) 3389 (m, NH₂), 3294 (s, NH₂), 1607 (vs. C=N), 1510 (s, ring), 1450 cm⁻¹ (s, ring); NMR (CF₃COOH) δ 9.50 (br, 1H, SH), 7.18 (m, 5H, C_6H_4 and CH=N), 4.03 (s, 3H, OCH_3), 3.07 (m, 9H, C-1,3,5,3',5' \underline{H}_2), 2.27 (m, 4H, C-4,4' \underline{H}_2); NMR (CD₃SOCD₃) δ 10.81 (br, ca. 2H, N $\underline{\text{H}}_2$), 8.97 (br, ca. 1H, SH); Found: C, 54.96; H, 5.70; N, 6.53; S, 29.14%. Calcd for C₂₀H₂₄N₂S₄O: C, 55.02; H, 5.54; N, 6.42; S, 29.36%.

3 - (3,4 - Methylenedioxyphenyl) - 1,3,4,5,6,7-hexahydro-1-thioxocyclopenta[d] [1,3]thiazine(IIIa). A solution of I (2 g), piperonal (2 g) and sulfuric acid (96%, 1g) in methanol (100 ml) was refluxed for 35 min. The resulting material was collected and recrystallized from N,N-dimethylformamide-water and dried at 110°C under reduced pressure for 1 day to give yellowish orange plates (IIIa): yield 3.6 g, 92%; mp 178— 179°C; UV $\lambda_{\max}^{\text{EtOH}}$ 341, 408 m μ (log $\varepsilon = 4.00$, 4.61); IR (CHCl₃) 3406 (s, NH), 1541 (s, conjd C=C), 1520 (s, ring), 1506 (vs, ring), 1446 cm^{-1} (s, ring); NMR (CF₃COOH) δ 9.42 (br, 1H, NH), 7.02 (m, 3H, C₆H₃), 6.14 (d, 1H, J=4 Hz, C-3 \underline{H}), 6.07 (s, 2H, OC \underline{H}_2 O), 3.05 (m, 4H, C-5,7 H₂). 2.37 (m, 2H, C-6 H₂); Found: C, 57.86; H, 4.46; N, 4.79; S. 22.08%. Calcd for $C_{14}H_{13}NS_2O_2$: C, 57.71; H, 4.50; N, 4.81; S, 21.99%.

3-(3-Methoxy-4-hydroxyphenyl)-1,3,4,5,6,7-hexahydro-1-thioxocyclopenta[d][1,3]thiazine (IIIb). A solution of I (2.5 g), vanillin (2.5 g) and sulfuric acid (96%, 1 g) in methanol (60 ml) was refluxed for 1 hr. The resulting material was worked up as above. Light lustrous orange prisms (IIIb): yield 4.1 g, 89%; mp 229—239°C with dec. (slow heating), 237—250°C with dec. (rapid heating); UV $\lambda_{\max}^{\text{EIOH}}$ 340, 408 m μ (log ε =3.93, 4.32); IR (CHCl₃) 3365 (s, NH), 1560 (s, conjd C=Cl), 1505 (vs. ring), 1444 cm⁻¹ (m, ring); NMR (CF₃COOH) δ 9.50 (br, 1H, NH), 7.12 (m, 3H,

C₉H₃), 6.15 (d, J=4 Hz, C-3 H), 3.97 (s, 3H, OCH₃), 3.00 (m, 4H, C-5,7 H₂), 2.36 (m, 2H, C-6 H₂); NMR (CD₃SOCD₃) δ 9.60 (s, 1H, OH), 9.37 (br, 1H, NH); Found: C, 57.31; H, 5.09; N, 4.80; S, 21.60%. Calcd for C₁₄H₁₅NS₂O₂: C, 57.32; H, 5.16; N, 4.78; S, 21.85%.

3 - (4 - Methoxyphenyl) - 1,3,4,5,6,7 - hexahydro-1-thioxocyclopenta[d][1,3]thiazine (IIIc). To a hot solution of I (2.5 g) and anisaldehyde (2.5 g) in methanol (100 ml), was added sulfuric acid (96%, 1 g) and refluxed for 30 min. The resulting mixture was worked up as above. Red long prisms (IIIc): yield 3.5 g, 81%; mp 179—180°C; UV $\lambda_{\max}^{\text{BioH}}$ 341, 408 m μ (log ε =3.70, 4.08): IR (CHCl₃) 3410 (s, NH), 1545 (s, conjd C=C), 1515 cm⁻¹ (vs. ring): NMR (CF₃COOH) δ 9.39 (br, 1H, N<u>H</u>), 7.34 (q, 4H, C₆H₄), 6.17 (d, 1H, J=4 Hz, C-3 H), 4.08 (s, 3H, OCH₃), 3.03 (m, 4H, C-5, 7 H₂), 2.38 (m, 2H, C-6 H₂); Found: C, 60.66; H, 5.34; N, 5.15; S, 23.13%. Calcd for C₁₄H₁₅NS₂O: C, 60.62; H, 5.45; N, 5.05; S, 23.11%.

3-Methyl-1,3,4,5,6,7-hexahydro-1-thioxocyclopenta [d][1,3]thiazine (IIId). A solution of I (5 g), acetaldehyde (5 g) and sulfuric acid (96%, 2 g) in methanol (150 ml) was allowed to stand at room temperature for 64 hr. The resulting reaction mixture was concentrated under reduced pressure and then water was added. The purple material which separated from the solution was recrystallized from methanolwater. Light orange plates (IIId): yield 4.5 g, 78%; mp 128—129°C; UV $\lambda_{\max}^{\text{EIOH}}$ 337, 405 m μ (log ε =3.72, 4.26); IR (CHCl₃), 3405 (s, NH), 1550 (vs. conjd C=C), 1515 cm⁻¹ (vs. ring); NMR (CF₃COOH) δ 9.40 (br, 1H, N $\underline{\text{H}}$), 5.17 (m, 1H, C-3 $\underline{\text{H}}$), 2.92 (m, 4H, C-5,7 $\underline{\text{H}}_2$), 2.25 (m, 2H, C-6 $\underline{\text{H}}_3$), 1.85 (d, 3H, C $\underline{\text{H}}_3$); Found: C, 51.83; H, 5.83; N, 7.62; S, 34.87%. Calcd for C₈H₁₁NS₂: C, 51.85; H, 5.99; N, 7.55; S, 34.61%.

3-Methyl-3-phenyl-1,3,4,5,6,7-hexahydro-1-thioxocyclopenta[d][1,3]thiazine (IIIe). A mixture of I (5 g), acetophenone (5 g), sulfuric acid (96%, 2 g) and methanol (150 ml) was refluxed for 3 hr and the reaction mixture was worked up as above. Light orange plates (IIIe): yield 1.6 g, 19.5%; mp 183—184°C (remelted at 195—197°C); UV $\lambda_{\max}^{\text{stot}}$ 340, 407 m μ (log ε =3.70, 4.30); IR (CHCl₃) 3400 (m, NH), 1550 (vs. conjd C=C), 1522 cm⁻¹ (s, ring); NMR (CF₃-COOH) δ 9.66 (br, 1H, N $\underline{\text{H}}$), 7.51 (s, 5H, C₆ $\underline{\text{H}}_5$), 3.00 (m, 4H, C-5,7 H₂), 2.31 (m, 2H, C-6 $\underline{\text{H}}_2$), 2.28 (s, 3H, C $\underline{\text{H}}_3$); Found: C, 64.41; H, 5.77; N, 5.40; S, 24.73%. Calcd for C₁₄H₁₅NS₂: C, 64.32; H, 5.79; N, 5.36; S, 24.53%.

3 - (2 - Hydroxyphenyl) - 1,3,4,5,6,7 - hexahydro - 1thioxocyclopenta[d][1,3]thiazine (IIIf). A solution of I (2 g), salicylaldehyde (2 g) and sulfuric acid (96%, 1 g) in methanol (70 ml) was refluxed for 10 min and then allowed to stand at room temperature for 1 hr. The reaction mixture was concentrated under reduced pressure, then water was added. The resulting red material was recrystallized from methanol-water. Red long plates (IIIf): yield 2.2 g, 67%; mp 201—215°C with dec. (slow heating), 211-227°C with dec. (rapid heating); UV λ_{mex}^{EtoH} 341, 408 m μ (log ε =3.87, 4.20); IR (CHCl₃) 3260 (s, OH), 3160 (s, br, NH), 1560 (s, conjd C=C), 1495 cm⁻¹ (vs. ring); NMR (CF₃COOH) δ 9.43 (s, br, 2H, OH and NH), 7.28 (m, 4H, C₆H₄), 6.58 (d, 1H, J=5 Hz, C-3 $\underline{\text{H}}$), 3.02 (m, 4H, C-5,7 $\overline{\text{H}_2}$), 2.35 (m, 2H, C-6 H₂); Found: C, 59.46; H, 4.92;

N, 5.25; S, 23.95%. Calcd for $C_{13}H_{13}NS_2O$: C, 59.28, H, 4.98; N, 5.32; S, 24.34%.

3-Phenyl-1,3,4,5,6,7-hexahydro-1- thioxocyclopenta[d][1,3]thiazine (IIIg). A solution of I (5 g), benzaldehyde (5 g) and sulfuric acid (96%, 2 g) in methanol (150 ml) was refluxed for 2 hr. The resulting material was recrystallized from methanol. Red prisms (IIIg): yield 5.1 g, 66%; mp 199—200°C; UV $_{\rm max}^{\rm EOOH}$ 340, 408 mμ (log ε=3.86, 4.28); IR (CHCl₃) 3400 (s, NH), 1546 (vs. conjd C=C), 1515 cm⁻¹ (s, ring); NMR (CF₃-COOH) δ 9.45 (br, 1H, NH), 7.49 (m, 5H, C₆H₅), 6.13 (d, 1H, J=4 Hz, C-3 H_1), 2.97 (m, 4H, C-5, H_2), 2.31 (m, 2H, C-6 H_2); Found: C, 63.06; H, 5.11; N, 5.70; S, 25.79%. Calcd for C₁₃H₁₃NS₂: C, 63.12; H, 5.30; N, 5.66; S, 25.92%.

3 - (2 - Methoxypehnyl) - 1,3,4,5,6,7 - hexahydro-1-thioxocyclopenta [d] [1,3] thiazine (IIIh). A mixture of I (5 g), o-methoxybenzaldehyde (5 g), sulfuric acid (96%, 2 g) and methanol (150 ml) was refluxed for 30 min and worked up as above. Light thick plates (IIIh): yield 5.3 g, 61%; mp 201°C; UV $\lambda_{\max}^{\text{EiOH}}$ 341, 408 mμ (log ε=3.88, 4.29); IR (CHCl₃) 3410 (s, NH), 1547 (vs. conjd C=C), 1510 cm⁻¹ (s, ring); NMR (CF₃COOH) δ 9.35 (br, 1H, NH), 7.28 (m, 4H, C₆H₄), 6.56 (d, 1H, J=6 Hz, C-3 H), 3.97 (s, 3H, OCH₃), 3.00 (m, 4H, C-5,7 H₂), 2.36 (m, 2H, C-6 H₂); Found: C, 60.73; H, 5.57; N, 5.10; S, 23.10%. Calcd for C₁₄H₁₅NS₂O: C, 60.62; H, 5.45; N, 5.05; S, 23.11%

3-(4-Hydroxyphenyl) - 1,3,4,5,6,7-hexahydro-1-thioxocyclopenta[d][1,3]thiazine (IIIi). A mixture of I (5 g), p-hydroxybenzaldehyde (5 g), sulfuric acid (96%, 2 g) and methanol (150 ml) was allowed to stand at room temperature for 30 min and worked up as above. Orange prisms (IIIi): yield 6.4 g, 77%; mp, 197—213°C with dec. (slow heating), 211—220°C with dec. (rapid heating); UV $\lambda_{\max}^{\text{EIOH}}$ 340, 408 mμ (log ε=3.88, 4.28); IR (CHCl₃) 3268 (s, OH), 3100 (m, br, NH), 1566 (s, conjd C=C), 1512 cm⁻¹ (vs. ring); NMR (CF₃COOH) δ 9.49 (br, 1H, NH), 7.27 (q, 4H, C₆H₄), 6.18 (d, 1H, J=4 Hz, C-3 H̄), 3.00 (m, 4H, C-5,7 H̄₂), 2.31 (m, 2H, C-6 H̄₂); Found: C, 59.28; H, 4.83; N, 5.33; S, 24.37%. Calcd for C₁₃H₁₃NS₂O: C, 59.28; H, 4.98; N, 5.32; S, 24.34%.

3 - (p-Tolyl) - 1,3,4,5,6,7-hexahydro-1-thioxocyclopenta [d] [1,3]thiazine(III). A mixture of I (3 g), ρ -tolualdehyde (3 g), sulfuric acid (96%, 1 g) and methanol (100 ml) was refluxed for 30 min and then water was added. The solution was allowed to stand

at room temperature for 12 hr. The resulting crystalline material was recrystallized from methanol and dried at 110°C under reduced pressure. Light orange prisms (IIIj): yield 2.3 g, 44%; mp 176—177°C; UV $\lambda_{\rm max}^{\rm EIGH}$ 340, 408 m μ (log ε =3.91, 4.34); IR (CHCl₃) 3420 (s, NH), 1545 (s, conjd C=C), 1515 cm⁻¹ (vs. ring); NMR (CF₃COOH) δ 9.50 (br, 1H, N $\underline{\rm H}$), 7.40 (s, 4H, C-5,7 $\underline{\rm H}_2$), 2.42 (s, 3H, C $\underline{\rm H}_3$), 2.38 (m, 2H, C-6 $\underline{\rm H}_2$); Found: C, 64.58; H, 6.05; N, 5.36; S, 24.23%. Calcd. for C₄H₁₅NS₂: C, 64.32; H, 5.79; N, 5.36; S, 24.53%.

3-Ethyl-3-methyl-1,3,4,5,6,7-hexahydro-1-thioxocyclopenta[d][1,3]thiazine(IIIk). A mixture of I (5 g), methyl ethyl ketone (22 g), morpholine (15 g) and ethanol (100 ml) was refluxed for 10 hr. The white precipitates which separated from the solution were filtrated, the filtrate was concentrated under reduced pressure and then water was added. The resulting yellow material was recrystallized from ethanolwater and dried at 138°C under reduced pressure for 1 day. Yellow plates (IIIk): yield 4.4 g, 67%; mp 173—174°C; UV $\lambda_{\text{max}}^{\text{EtOH}}$ 336, 406 m μ (log ε =3.68, 4.25); IR (KBr) 3180 (s, NH), 1565 (m, conjd C=C), 1520 cm⁻¹ (vs. ring); NMR (CD₃COCD₃) δ 9.40 (br, 1H, NH), 3.00 (m, 4H, C-5,7 \underline{H}_2), 2.25 (m, 4H, C-6 \underline{H}_2) and CH_2CH_3), 1.85(s, 3H, CH_3). 1.15(t, 3H, CH_2CH_3); Found: C, 56.53; H, 7.00; N, 6.54; S, 29.99%. Calcd for $C_{10}H_{15}NS_2$: C, 56.32; H, 7.09; N, 6.57; S, 30.01%.

Compound IIIk was also obtained by refluxing a mixture of I (5 g), methyl ethyl ketone (5 g), sulfuric acid (96%, 2 g) and methanol (100 ml) for 1 hr; the yield was 57%.

3,3 - Dimethyl - 1,3,4,5,6,7 - hexahydro - 1 - thioxocyclopenta [d] [1,3]thiazine (IIII). A mixture of I (2.5 g), acetone (3 g), sulfuric acid (96%, 1 g) and methanol (100 ml) was refluxed for 1.5 hr and then worked up as above: yield 51%, mp 172—173°C, undepressed on admixture with an authentic specimen. IR and UV absorption spectra of IIII were also identical with those of authentic specimen.²⁾

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